

SOME RESULTS REGARDING THE ULTRASONIC CAVITATION

by
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Abstract: The aim of this paper is to prove the existence of the material erosion in ultrasonic field and to measure the intensity of the cavitation phenomenon. We give a mathematical model for the voltage induced by the cavitation.

Keywords: ultrasonic cavitation, erosion, voltage

1. PRELIMINARIES

When an acoustic wave propagates through a liquid containing microscopic gas inclusions, these "nucleation sites" can be mechanically activated, at which point they spawn free bubbles which then undergo highly energetic volume pulsations. Associated with these pulsations is a broad range of linear and nonlinear mechanical behavior, the nature of which will primarily depend on the acoustic pressure amplitude and the equilibrium bubble. This activity, which is termed acoustic cavitation, is often accompanied with other physical (erosion, unpassivation and emulsification) or chemical (the production of radical and excited species; single electron transfer) interactions.

We chose to study the mechanical erosion in the ultrasonic field and some mathematical aspects related to the electric field produced by the cavitation bubbles. Rayleigh gave a mechanical explanation of the materials destruction by cavitation. Other scientists consider that these damages are produced by chemical or electrical effects; but, it seems that they are the results of the mechanical pressure inside the bubble.

Plesset and Elis proved that the plastic deformation appears everywhere where a cavitation exists. Their studies bring to the conclusion that the chemical action can't be the primary cause of the erosion.

In order to determine the effect of the ultrasound on the propagation medium, the spectral analysis was made, before and after ultrasonic treatment. It was also weighted material put in the sonication bath and it was determined the intensity of the ultrasonic action on the propagation medium.

The liquid used was the water.

2. EXPERIMENTAL RESULTS

We used an ultrasound generator of the tip I.U.S. - 150 and of maximum power 180W, with three degrees of electrical power: I: 80W, II: 120W, III: 180W.

In the sonication bath was introduced, in water, an aluminium foil, weighed before and after the experiment. The results obtained after 30 minutes, for each power are given in the table 1, where:

- m_i is the initial mass of the aluminium foil (g),
- m_j is the mass of the aluminium foil, after the sonication (g),
- τ is the sonication time, in minutes (in our case, 30 minutes),
- $\Delta m = m_j - m_i$,
- s_0 is the foil surface (cm²).

It can be seen that, if the power is greater, the loss of the mass, also increases. For the second case, the mass variation is 3.83 times greater than in the first case and in the third case, the lost mass is 5.38 times greater than in the first case.

Table 1. The aluminium erosion

Power	m_i	m_j	$\frac{\Delta m}{2s_0\tau}$	$\frac{\Delta m}{m_i}$
80	0,1302	0,1169	0.0000410	0,1021505
120	0,1169	0,0748	0.0001299	0,3233486
180	0,0748	0,1791	0.0001304	0,5499232

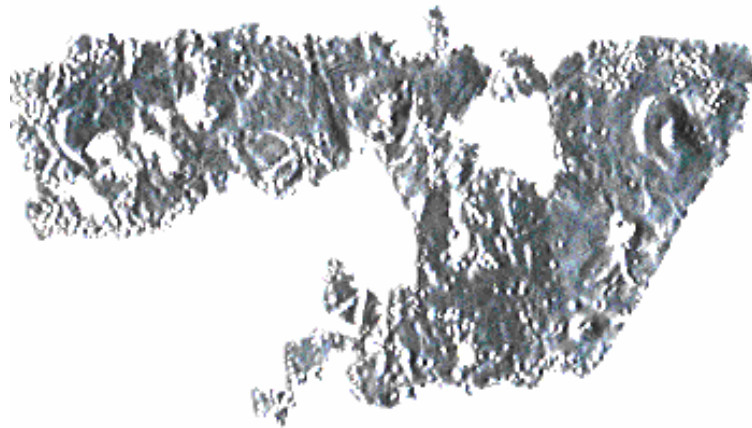


Figure1. The aluminium erosion

In order to determine the caloric power we have following formula is used:

$$P_c = \frac{Q}{\tau} \quad (1)$$

where:

- τ is the time of the ultrasounds application,
- Q is the quantity of the heat obtained by the ultrasounds propagation in the studied liquid:

$$Q = C(\theta - \theta_0), \quad (2)$$

- θ is the equilibrium temperature in the calorimeter,
- θ_0 is the initial temperature of the calorimeter,
- C is the caloric capacity of the calorimeter.

We introduced in the calorimeter a piece of copper, with the mass $m = 0.1027\text{Kg}$, with the specific heat $c = 393.86\text{J/KgK}$ and the initial temperature $\theta = 100^\circ\text{C}$. The final temperature was $\theta_1 = 33.3^\circ\text{C}$.

The quantity of the waste heat by the copper is equal with absorbed heat by the calorimeter if the calorimeter contains the same water quantity as during the measurement of the caloric power given by the ultrasounds. Thus:

$$C(\theta - \theta_0) = m_{Cu}c_{Cu}(100 - \theta) \Rightarrow \quad (3)$$

$$C = \frac{0.0127 \cdot 393.86 \cdot (100 - 33.3)}{33.3 - 22.2} = 243 \frac{\text{J}}{\text{K}}$$

Using the ultrasonic generator of 1MHz, we measured the temperature increasing of the ultrasonically treated liquid. We've determined the temperature variations and the energy values for the different power degrees (table 2).

Table 2. The energy values for different power degree

Power (W)	θ_i ($^\circ\text{C}$)	θ_f ($^\circ\text{C}$)	$\Delta\theta$ ($^\circ\text{C}$)	Q_c (KJ)	Q_a (KJ)
67	20.6	30.2	9.6	319.680	219.33
134	22	34.5	12.5	639.360	438.66

200	22.2	40.1	17.9	959.040	658.00
267	23.3	42.2	22.9	1.278.720	877.333
333	24.4	53.4	29	1.598.400	1.096.66
400	27.8	58.1	30.3	1.918.080	1.316.03

The data from the table 2, the columns 5 and 6 are calculated with the relations:

$$Q_c = P_c \tau, Q_a = P_a \tau, \quad (5)$$

where:

τ is the time of the ultrasounds application,

Q_c is the caloric power,

Q_a is the acoustic power, given by:

$$P_a = \frac{2\pi m v^2 A^2}{\tau} = \frac{2\pi \rho V v^2 A^2}{\tau}, \quad (6)$$

In the formula (6), the symbols have the following meanings:

- m - the mass of the reaction liquid (in our case, the water),

- v - the ultrasonic frequency,

- A - the amplitude of the ultrasonic oscillation.

Making a comparison between the binding energy for some chemical compounds, it results that the total energy, received by ultrasonic treatment is greater than the binding energy and some chemical compounds can be broken.

Table 3. The binding energy for some chemical compounds

Type of chemical compounds	Binding energy (Kj/mol)
O-H	428
N=N	409
C-S	259
C-C	518
C-N	305
C-Cl	338
N-O	157
C-Br	276

C-OH	360
C-CH ₃	348
C-I	238

3. MATHEMATICAL MODEL

In order to study the voltage appeared in the ultrasonic cavitation, we used the appliance drawn in the figure 2. The data obtained, using the acquisition card, at the first degree of power of the ultrasound generator, are represented in the figure 3.

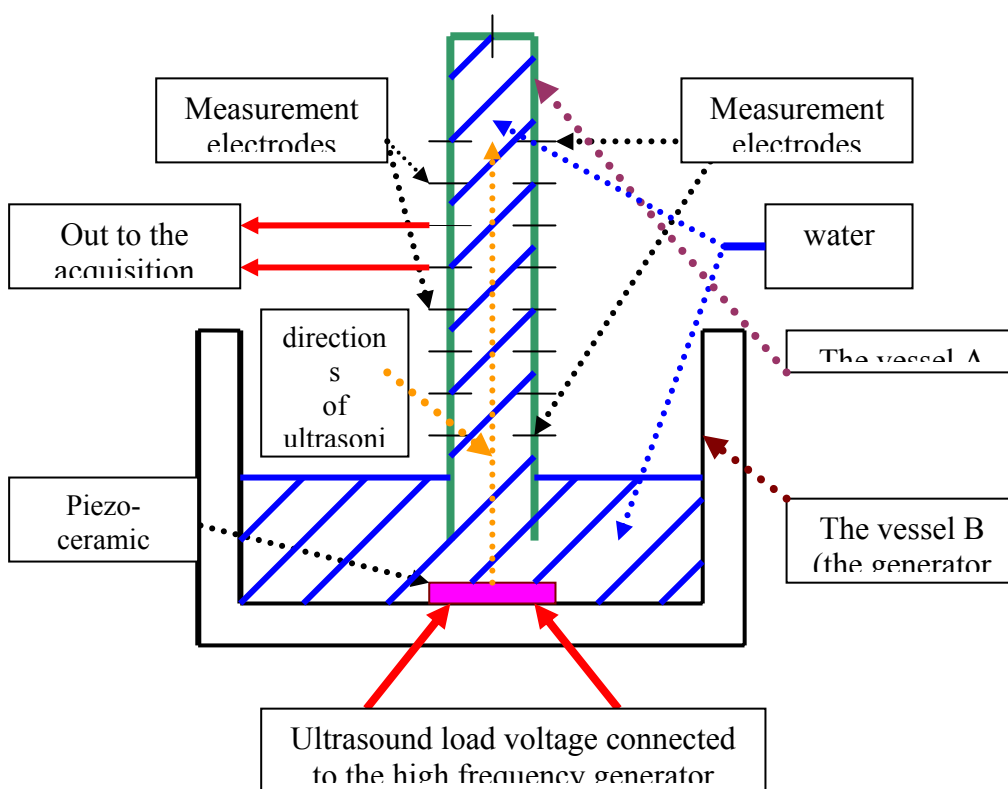


Figure 2. Experimental appliance

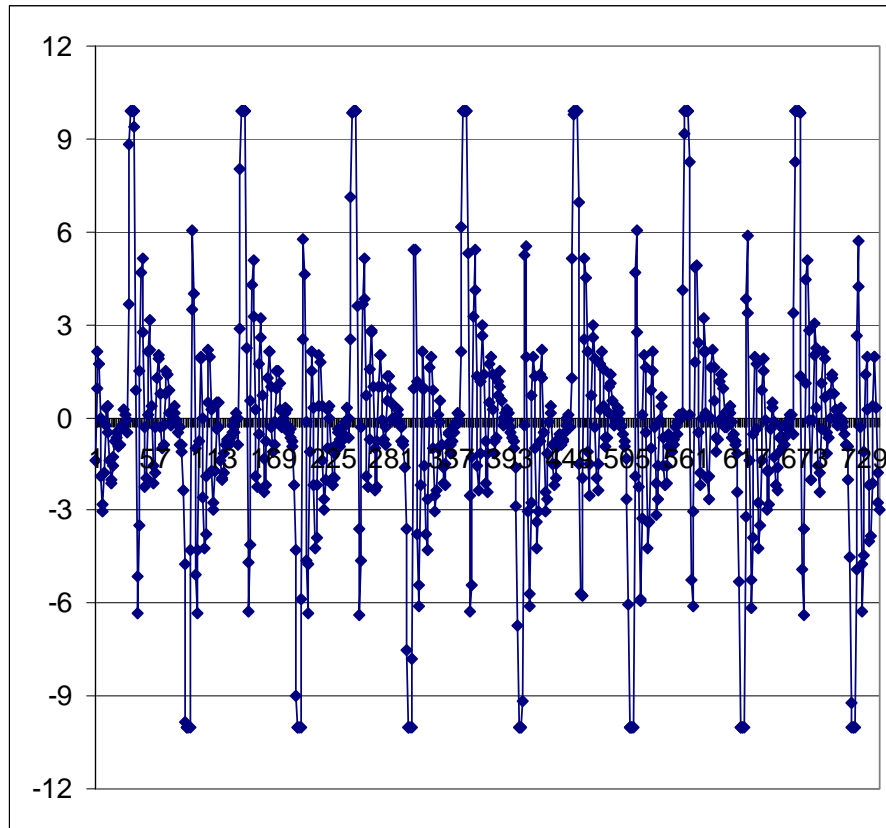


Figure 3. The signal form

It can be seen that there exists a periodicity of the signal and after the study of the given values, the determined period was 105. The study was made only for a period using the Box - Jenkins method.

The model proposed is an ARIMA(2, 0, 0), without a constant term:

$$V_n = 1.5636298V_{n-1} - 0.8913194V_{n-2} + \varepsilon_n, n \in \mathbf{N}, n \geq 3, \quad (7)$$

where ε_n is the residual, which is a white noise.

To prove the fact that the model is a good one we studied the autocorrelation function and the partial autocorrelation function for the residuals. The values of these functions are inside the confidence intervals (at 0.95 confidence level) (The figures 4 and 5)

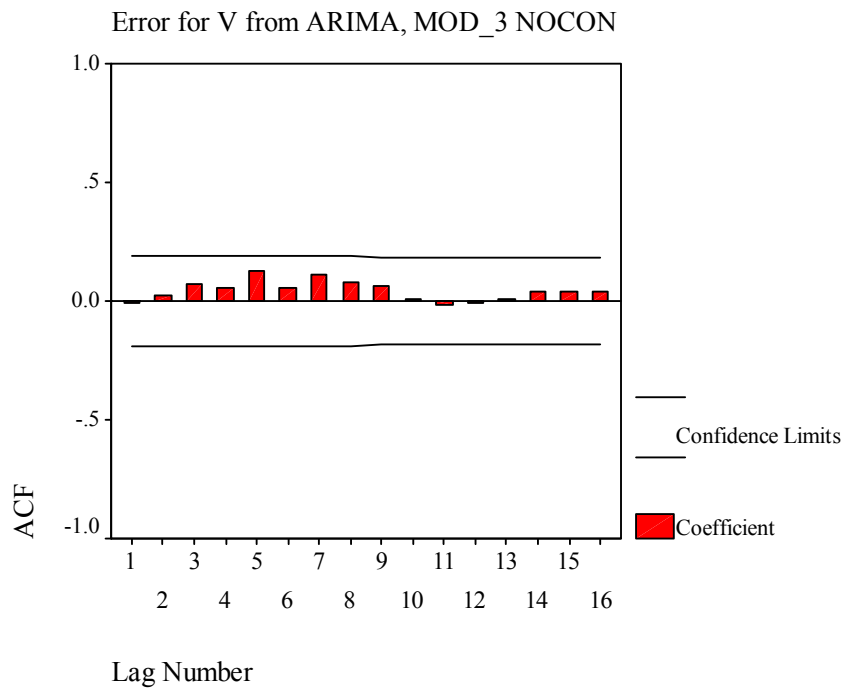


Figure 4. The autocorrelation function for the residual in the model (7)

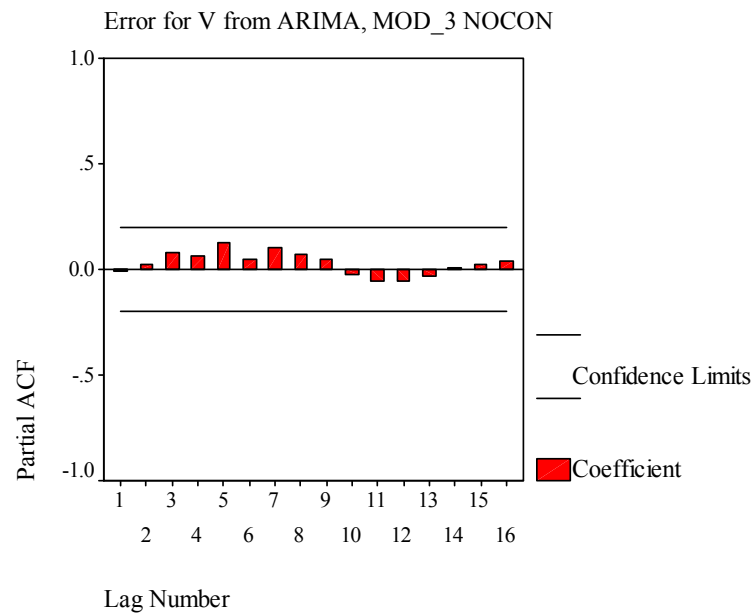


Figure 5. The partial autocorrelation function for the residual in the model (7)

The values of the Box –Ljung statistics are in the interval $[0.008, 6.234]$, so they are less than $\chi^2(100)$. The probabilities of the acceptance of the white noise hypothesis for the residuals are big, so the residuals form a white noise and the model is well selected.

3. CONCLUSIONS

The ultrasonic treatment applied on some material produces them erosion. Also, the cavitation bubbles are electric charged and the existence of that tension was proved using the experimental appliance. The mathematical form of the voltage induced by the cavitation was also determined. In our study (to appear) we saw that this form is different for the different types of liquid.

References

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